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VERIFICATION OF A TRANSLATION

I, the below named translator, hereby declare that:

My name and post office address are as stated below;

That I am knowledgeable in the English language and in the language in which the below identified application was filed, and that I believe the English translation of International Application No. PCT/JP2004/009143 is a true and complete translation of the above-identified International Application as filed.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

Dated this 14th day of December 2005.

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SPECIFICATION

PRINTING PAPERS

TECHNICAL FIELD

[0001]

The present invention relates to bulky printing papers with reduced loss of brightness, i.e. reduced discoloration, especially printing papers well-suited for books.

PRIOR ART

10 [0002]

An effective way to utilize papermaking pulp derived from forest resources (in view of the growing concerns of environmentalists), is to prepare paper containing the highest possible proportions of mechanical pulp

- 15 (hereinafter referred to as MP) having a high yield of about 90% or more, such as stone groundwood pulp, refined ground wood pulp, pressurized groundwood pulp, thermomechanical pulp, chemithermomechanical pulp and bleached chemithermomechanical pulp rather than chemical pulp (hereinafter referred to as CP) having a process yield of about 50% such as kraft pulp, alkaline pulp and sulfite pulp. It is also desirable to improve the utilization ratio of deinked pulp (hereinafter referred to as DIP) made
- 25 [0003]

On the other hand, weight reduction is a major quality requirement for paper. The weight reduction of paper here means weight reduction of paper retaining the

by defiberizing paper once used.

thickness, i.e. low-density (bulky) paper. To prepare such low-density (bulky) paper, MPs having stiffer fibers such as groundwood pulps obtained by grinding wood with grinders or thermomechanical pulps obtained by fibrillating wood with refiners are also more effective for density reduction 5 than CPs prepared by extracting lignin (a reinforcing component in fibers with chemicals); especially, ground wood pulps greatly contribute to density reduction. Thus. MPs are more preferred for an effective utilization of 10 resources and achieving weight reduction of paper, but there is a serious problem of change in brightness over a period of time, i.e., the problem of discoloration, especially in high-brightness papers. Deinked pulps also help save wood resources by increasing their proportions, 15 but they show a similar discoloration quality to that of MPs because waste papers are pulped without separating MPs and there are few cases in which they are definitely sorted by paper quality into woodfree papers, newspapers, magazines, dodgers, coated papers, etc. before they are 20 pulped. For this reason, carefully sorted waste papers should be used to include deinked pulps into woodfree papers. Thus, it is favorable to include the highest possible proportions of MP or DIP in view of the resources problem, and the inclusion of MP would also contribute to 25 weight reduction of paper, but involves the problem of discoloration over a period of time, which is a serious problem especially in the case of paper used in books that are stored and displayed at bookstores for a long period.

[0004]

Discoloration of pulp over a period of time is a common issue for any type of pulp including CP, semichemical pulp (hereinafter referred to as SCP), MP and DIP.

The extent of discoloration is mainly because of the presence of residual amounts of lignin or modified lignin in pulp, and the discoloration tends to advance as the residual amount of lignin or modified lignin increases. Thus, discoloration takes place most rapidly in MP

containing large amounts of residual lignin or modified lignin, followed by SCP, and then CP. Discoloration in DIP is greatly influenced by the MP content or other factors.

[0005]

During bleaching of pulp, lignin is oxidized to 15 produce a hydroquinone as modified lignin. hydroquinone is readily oxidized via a reaction shown in Scheme 1 to give a quinone, resulting in coloration. lignin remaining unoxidized in the pulp is also excited by UV light and oxidatively degraded via the reaction shown in 20 Scheme 2 to freshly produce a quinone compound, resulting in coloration. Thus, conventional methods for improving discoloration in pulp normally comprise degrading or removing causative agents of discoloration such as lignin or modified lignin by adding a bleaching chemical or 25 alkaline chemical during the pulp bleaching process. [0006]

The mechanism of discoloration in MP is the same as that of the lignin-induced discoloration in pulp described

above, but discoloration in MP is more severe than in CP and SCP because of the presence of higher residual amount of lignin and modified lignin than in CP and SCP. cause of the strong coloration in MP is that lignin is oxidized during hydrogen peroxide-bleaching to produce a 5 hydroquinone, which is readily oxidized into a quinone. Discoloration becomes more severe in MP having higher brightness because more hydroquinone is produced at higher degrees of bleaching. The lignin remaining unoxidized in 10 MP is also excited by UV light and oxidatively degraded to newly produce a quinone compound, resulting in coloration. Thus, the main factor behind severe discoloration in MP is the quinone compound, and if this quinone compound could be degraded in advance, the discoloration in MP would be greatly reduced and many benefits can be expected such as 15 (1) quality stabilization of current paper products containing MP, (2) enrichment in MP in current paper products with limited content of MP due to the discoloration in MP, thereby providing bulky and light 20 weight paper products, (3) the possibility of the development of new products containing MP, etc. [0007]

In order to solve this problem of discoloration in MP, many proposals have been made, and a method using a water
25 soluble UV absorbent and a photostabilizer in combination has been recently proposed (see non-patent document 1). A method in which the aromatic ring contained in lignin in MP is reduced was also shown (see non-patent document 2).

However, UV absorbents or the like have the disadvantage that their effect does not last for a long period because they are also lost because of UV light. On the other hand, when a rhodium catalyst was used for reducing the lignin aromatic ring, it took a long period, such as 5 days to partially hydrogenate the aromatic ring in the hydrogenation reaction of the aromatic ring of the lignin isolated from wood at room temperature in an aqueous alcohol solution, and this catalyst seems very difficult to directly react with the lignin present in pulp fibers because it is used as an emulsion. Thus, all of these conventional methods have disadvantages such as insufficient anti-fading effect, long processing time, lack of economy or practicability, etc. at present.

15 [0008]

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Discoloration also occurs in CP though it is less serious than in MP. As in MP if discoloration in CP could also be reduced, such benefits as quality stabilization of CP-based paper products and the possibility of the development of new products would become possible.

[0009] Scheme 1

[0010] Scheme 2

5 A conventional method for improving discoloration in CP was proposed in which causative agents of discoloration such as lignin or hexenuronic acid are degraded / removed with chemicals or the like. In this method, however, large amounts of chemicals or the like must be used to 10 degrade/remove lignin and these treatments cause problems such as yield loss in bleached pulp or strength loss in pulp fibers. Large amounts of acids are required to remove hexenuronic acid, which also invites the problem of strength loss in pulp fibers. If the quinone compound 15 could be degraded in advance, such benefits as quality stabilization of CP-based paper products and the possibility of the development of new products would be expected.

[0011]

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As an example of prior art, a document shows that discoloration is improved better at higher temperatures with higher hydrogen peroxide concentrations being maintained for longer reaction periods during the hydrogen peroxide-bleaching stage under alkaline conditions (see non-patent document 3). A bleaching process has been patented, characterized in that an alkali is added latterly in the first reaction stage of successive chlorine dioxide stages when a pulp derived from a lignocellulose material is bleached according to a bleaching sequence consisting of chlorine and/or chlorine dioxide stage - alkali/oxygen stage -chlorine dioxide stage -chlorine dioxide stage and wherein, the ratio of chlorine dioxide at the first to the second stages is 40/60 to 70/30 (see patent document 1). A multi-stage bleaching process has also been patented, comprising the steps of treating a pulp derived from a lignocellulose material with chlorine, alkaline extraction, and then a hypochlorite, characterized in that 1.0% or more of an alkali is added on the basis of the bone dry weight of the pulp during the hypochlorite-bleaching stage and the pulp is treated at a temperature of 60°C or more (see patent document 2). A technique has been disclosed, which is characterized in that a pulp bleached in a sequence including at least one chlorine bleaching stage is treated with xylanase and further bleached in a bleaching sequence including a hypochlorite stage and a chlorine dioxide stage (see patent document 3). A process for preparing a pulp with high degree of brightness has been disclosed,

comprising further bleaching a bleached pulp derived from a lignocellulose material through a sequence of a hyposulfite bleaching stage under high temperature and high alkali conditions and a chlorine dioxide bleaching stage, wherein the second chlorine dioxide bleaching stage is performed at a chlorine dioxide concentration of 1% by weight to 3% by weight on the basis of the bone dry weight of the pulp in a chlorine dioxide/alkali ratio of 1/0.05 to 1/0.3 and at a high temperature (see patent document 4).

10 [0012]

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It has become known from recent new findings that not only conventional lignin and modified lignin but also hexenuronic acid are responsible for discoloration in ECF or TCF bleached pulp. This hexenuronic acid is produced by demethylation from methylglucuronic acid in hemicellulose during the cooking step. This hexenuronic acid is said to be responsible for discoloration in pulp. An acid treatment at relatively high temperatures was proposed to remove the hexenuronic acid. According to this method, the hexenuronic acid and modified lignin are removed by acid hydrolysis by treating unbleached pulp at high temperatures under acid conditions. For example, a technique has been disclosed wherein a suspension of a cellulose pulp prepared by the sulfate process or alkaline process is heated and treated at about 85 to 150°C and a pH of about 2 to 5 to remove at least about 50% of hexenuronic acid in the cellulose pulp, thereby decreasing the Kappa number of the pulp by 2 to 9 units (see patent document 5).

[0013]

One of the applicants of the present invention, National Institute of Advanced Industrial Science and Technology, already filed three patent applications 5 relating to pulp bleaching methods involving irradiation with UV and/or visible light. Specifically, a technique is disclosed involving irradiation with UV and/or visible light in bleaching pulp using a reducing agent (see patent document 6). Another technique is disclosed involving 10 irradiation with UV and/or visible light in the presence of an organic peroxide represented by ROOR' as an oxidizing agent (see patent document 7). Still another technique is disclosed combining the use of hydrogen peroxide and UV and/or visible light (see patent document 8). The present 15 invention relates to the making of printing papers using a pulp with improved discoloration by applying the techniques of the inventions of these prior applications to bleached pulp.

[References]

20 Non-patent document 1: Yuan, Z., et al., J. Pulp Paper Sci., 28 (5), 159 (2002).

Non-patent document 2: Hu, T. Q., et al., J. Pulp Paper Sci., 25(9), 312 (1999).

Non-patent document 3: The Bleaching of Pulp, P382, 25 P384, TAPPI PRESS (1979), N. Hartler, TAPPI 43 (11) 903 (1960).

Patent document 1: JPB No. Hei 7-6147.

Patent document 2: Japanese Patent No. 2115315.

Patent document 3: JPA No. Hei 6-101185.

Patent document 4: JPA No. Hei 9-105091.

Patent document 5: JPA No. Hei 10-508346.

Patent document 6: Japanese Patent No. 3371228.

Patent document 7: Japanese Patent No. 3368336

Patent document 8: JPA No. 2002-88673.

DISCLOSURE OF THE INVENTION PROBLEMS TO BE SOLVED BY THE INVENTION

10 [0014]

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An object of the present invention is to obtain printing papers with reduced discoloration by using a pulp with improved discoloration, thereby providing printing papers with remarkably improved discoloration characterized by (1) reduced brightness change, i.e. reduced discoloration; (2) weight reduction especially by including a larger quantity of MP; (3) achieving resource conservation and environmental friendliness by including higher proportions of MP or DIP, etc..

20 MEANS FOR SOLVING THE PROBLEMS

[0015]

As a result of careful studies, we found that printing papers with remarkably improved discoloration can be obtained by including a pulp consisting of one member or a mixed pulp of two or more members of bleached or unbleached mechanical pulps, semichemical pulps, chemical pulps and deinked pulps irradiated with UV and/or visible light in the presence of at least one additive selected

from the group consisting of reducing agents, oxidizing agents and hydrogen-donating organic compounds. Printing papers with remarkably improved discoloration can also be obtained by irradiating, a printing paper or a precursor thereof in the production process, with UV and/or visible light in the presence of at least one additive selected from the group consisting of reducing agents, oxidizing agents and hydrogen-donating organic compounds.

EFFECT OF THE INVENTION

10 [0016]

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Discoloration of paper is greatly reduced by providing a method for improving discoloration comprising preparing a paper containing a pulp irradiated with UV / visible light in the presence of at least one compound selected from the group consisting of reducing agents, oxidizing agents and hydrogen-donating organic compounds or irradiating prepared paper with UV / visible light in the same manner, and especially such benefits as environmental friendliness are achieved because light and bulky papers can be prepared from smaller amounts of wood resources by treating MP having a high yield or MP-containing papers.

PREFERRED EMBODIMENTS OF THE INVENTION
[0017]

Pulps used in printing papers of the present

25 invention include raw pulps such as bleached or unbleached mechanical pulps, semichemical pulps, chemical pulps and deinked pulps derived from wood. Either one member or a mixture of two or more members of these pulps may be used.

The term "bleached" here means that the pulp has gone through e.g., a normal multi-stage bleaching process or a short sequence bleaching process in contrast to unbleached pulps. The present invention is preferably directed to bleached mechanical pulps because the invention works very effectively in the case of pulps showing significant discoloration.

[0018]

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Compounds that can be used to improve discoloration

of pulp used in the present invention include known
reducing agents, oxidizing agents and hydrogen-donating
organic compounds commonly used as bleaching/decoloring
agents. The pulps are treated in the presence of at least
one compound among these groups.

15 [0019]

Such reducing agents include, for example, sulfite or bisulfite ion, hydrosulfites, borohydride compounds, etc.

The borohydride compounds are normally represented by general formula (1) or (2) below.

20 [0020]

 $M (BR_{4-n}H_n)_m \qquad (1)$

wherein n = an integer from 1 to 4, m = an integer from 1 to 3, M = a metal ion, organic ion or inorganic ion, R = a hydrocarbon group or substituted hydrocarbon group or

25 BR_{3-n}H_n (2)

wherein n = an integer from 1 to 3, R = a hydrocarbon group or substituted hydrocarbon group.

In general formula (1) above, the metal ion includes

monovalent metal ions such as alkali metals, divalent metal ions such as alkali earth metals, and trivalent metal ions; and the organic ion may be any stable ion, especially a quaternary ammonium ion. Substituents for R include 5 aliphatic hydrocarbon groups preferably containing 1 to 20, more preferably 1 to 10 carbon atoms, aromatic hydrocarbon groups preferably 6 to 20, more preferably 6 to 14 carbon atoms, alkyl-substituted hydrocarbon groups preferably containing 7 to 40, more preferably 7 to 24 carbon atoms, 10 etc. When two or more substituents for R exist, R may be the same or different. Especially preferred borohydride compounds for use in the present invention are sodium borohydride or tetrabutylammonium borohydride. borohydride or tetrabutylammonium borohydride or mixtures 15 thereof can also be used.

Preferred oxidizing agents are peroxides. As for peroxides, both organic peroxides and inorganic peroxides can be used. Organic peroxides that can be used are compounds represented by general formula (3) below: [0022]

ROOR' (3)

[0021]

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wherein R and R' may be the same or different and represent a hydrocarbon group, alkylcarbonyl group, arylcarbonyl group, alkoxycarbonyl group, aryloxycarbonyl group, formyl group or hydrogen.

The hydrocarbon group includes aromatic hydrocarbon groups such as phenyl, naphthyl, biphenyl and anthryl

groups; aliphatic hydrocarbon groups; and substituted forms thereof; and the alkylcarbonyl group includes acetyl, ethylcarbonyl and propionylcarbonyl groups and substituted forms thereof. The arylcarbonyl group includes benzoyl,

- naphthylcarbonyl and biphenylcarbonyl groups and substituted forms thereof. The alkoxycarbonyl group includes methoxycarbonyl and ethoxycarbonyl groups and substituted forms thereof. The aryloxycarbonyl group includes phenoxycarbonyl, naphthoxycarbonyl and
- biphenyloxycarbonyl groups and substituted forms thereof.

 Specific examples of these organic peroxides include, for example, peracids such as perbenzoic acid and derivatives thereof, peracetic acid and performic acid; esters of these peracids; and percarbonic acid and esters thereof.
- Inorganic peroxides include hydrogen peroxide, sodium percarbonate, sodium peroxide, etc. Other oxidizing agents include those containing no halogens such as ozone and oxygen.

[0023]

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20 Preferred hydrogen-donating organic compounds are alcohols and amides, among which preferred alcohols are primary alcohols represented by general formula (4) below, specifically including ethyl alcohol, benzyl alcohol, furfuryl alcohol, etc. Preferred amides are formamide and acetamide. However, any compound having hydrogen-donating ability can be used as hydrogen donor organic compounds without being limited to the alcohols and amides mentioned above.

RCH₂OH (4)

wherein R represents a hydrocarbon group or a substituted hydrocarbon group.

[0024]

In the pulp treatment used for printing papers of the present invention, the reducing agents, oxidizing agents and hydrogen-donating organic compounds may be used alone in the absence of a solvent, but should desirably be used as a dispersion or solution in a solvent transparent to UV / visible light. Such solvents include water, alcohols, linear or cyclic alkanes, ethers, etc. as single solvents or mixed solvents thereof, preferably water.

[0025]

agents and hydrogen-donating organic compounds to be used is not specifically limited so far as it is at or below the saturated concentration of the compound in the solvent, but suitably 0.01 to 40% (weight/volume), more preferably 0.1 to 20% in the solvent. The amount of any of the reducing agents, oxidizing agents and hydrogen-donating organic compounds to be used is 0.05 to 50% solids by weight, more preferably 0.1 to 25% solids by weight on the basis of pulp solids.

[0026]

25 The UV and/or visible light is not specifically limited, but desirably used at a wavelength of about 180 to 800 nm, preferably 200 to 500 nm. This is because lignin, para-quinone and ortho-quinone have maximum absorption

wavelength of 280 nm, 360 nm and 390 to 410 nm,
respectively. Suitable light sources include common light
sources such as low-pressure mercury lamps, high-pressure
mercury lamps, xenon lamps and black lights; as well as
various excimer lamps and various lasers. For mass
processing, the common light sources mentioned above are
preferably used. In view of high-speed processing; it is
desirable to use laser light sources. The laser light
sources are not specifically limited and the laser light
may be pulsed or continuous, but it is preferable to use
excimer lasers (ArF excimer laser, KrF excimer laser, XeCl
excimer laser, XeF excimer laser, etc.), argon ion laser,
krypton ion laser, the second and third harmonics of YAG
lasers, etc.

15 [0027]

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The irradiation intensity is not specifically limited, but suitably 0.1 mJ/cm²/pulse to 1.0 kJ/cm²/pulse for pulsed radiation and 0.1 mW/cm² to 10 kW/cm² for continuous radiation. The irradiation temperature is not specifically limited either, but preferably -80 to 100°C, more preferably 0 to 80°C. The irradiation time is appropriately determined taking into account the masses of potential coloring materials contained in raw pulps, the types and concentrations of additives or solvents, and the type and intensity of the irradiating UV light and/or visible light, but normally 1 to 60 minutes will suffice. [0028]

The pulps used in printing papers of the present

invention can be irradiated with UV and/or visible light in the presence of at least one compound selected from reducing agents, oxidizing agents and hydrogen-donating organic compounds in any manner without limitation. 5 method according to a preferred embodiment comprises 1) providing a solvent which contains at least one compound selected from reducing agents, oxidizing agents and hydrogen-donating organic compounds and which is transparent to UV and/or visible light, 2) dispersing a raw 10 pulp in the solvent and then 3) irradiating the dispersion with UV and/or visible light. Specifically when the solvent is water, the aqueous suspension of pulp issuing from the bleaching step is combined / mixed with at least one compound selected from reducing agents, oxidizing 15 agents and hydrogen-donating organic compounds and then irradiated with UV and/or visible light. An alternative method comprises 1) forming a raw pulp into a sheet or a thin layer, 2) then impregnating it with a solvent containing at least one compound selected from reducing 20 agents, oxidizing agents and hydrogen-donating organic compounds, and then 3) irradiating it with UV and/or visible light. Specifically when the solvent is water, the aqueous suspension of pulp having gone through bleaching and issuing from the bleaching step is dewatered through an 25 apparatus such as a pulp sheet machine into a pulp sheet, which is then impregnated with at least one compound selected from reducing agents, oxidizing agents and hydrogen-donating organic compounds and irradiated with UV

and/or visible light.
[0029]

Another possible method comprises 1) preparing a printing paper in a paper machine using raw pulp and papermaking fillers and various papermaking chemicals, 2) then impregnating or coating it with a solvent containing at least one compound selected from reducing agents, oxidizing agents and hydrogen-donating organic compounds, and then 3) irradiating it with UV and/or visible light.

10 [0030]

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Although the anti-fading mechanism of pulp treated by the methods described above has not been known in detail, we assume the following. That is, potential coloring materials contained in pulp, such as lignin in MP, are oxidatively degraded by irradiating with UV and/or visible 15 light during bleaching to produce a hydroquinone. Unmodified lignin remaining unoxidized during bleaching produces various quinones. The present invention takes advantage of the phenomenon in which these quinones are further electronically excited from the ground state into 20 the excited state by UV and/or visible light, and as a result, converted into a more reactive state, thereby enhancing the reaction with the coexisting of at least one compound selected from reducing agents, oxidizing agents 25 and hydrogen-donating organic compounds to promote the degradation reaction of the coloring materials or the hydrogenation reaction of multiple bonds; or conversely, the phenomenon in which at least one compound selected from reducing agents, oxidizing agents and hydrogen-donating organic compounds is converted into a more active substance by irradiation with UV and/or visible light, thereby improving the reactivity with the coloring materials to enhance the degradation reaction of the coloring materials or the hydrogenation reaction of multiple bonds.

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The method previously proposed for reducing discoloration by incorporating an additive such as a thiol compound into paper made from MP had the problem of unpleasant odor and high toxicity in addition to the problem that it did not provide a radical improvement because it merely promoted the reaction of quinones into hydroquinones and caused discoloration again by UV light once the additives had been consumed. However, the present invention not only completely resolves all these problems by adopting a unique means, i.e. irradiation with UV and/or visible light in the presence of at least one compound selected from reducing agents, oxidizing agents and hydrogen-donating organic compounds, but also has a remarkable advantage due to environmental friendliness.

Another attempt to reduce discoloration by applying a UV absorbent on the paper surface was effective only on the surface, but not effective in reducing discoloration on areas where the UV absorbent was not applied; for e.g. the trimmed edges of the leaves of a book. In papers of the present invention, discoloration on the edges can also be

reduced by performing a discoloration-improving treatment on the whole sheet of paper or the whole of the pulp.
[0033]

The amount of the pulp having undergone the

discoloration-improving treatment to be included into
printing papers of the present invention is not
specifically limited, and the whole sheet of paper once
prepared may also undergo the discoloration-improving
treatment as described above. However, to increase

efficiency, it is more preferable to perform the
discoloration-improving treatment on only MP or DIP
containing MP because discoloration is more significant in
MP than CP.

[0034]

Printing papers of the present invention can contain paper strength enhancers. Exemplary paper strength enhancers include starches, modified starches, polyacrylamide, polyvinyl alcohol, polyamide-polyamine resins, urea-formalin resins, melamine-formalin resins, polyethylene imines, etc. The amount of the paper strength enhancers to be contained is preferably 0.1% by weight or more and 2% by weight or less on the basis of the bone dry weight of pulp.

[0035]

In addition to the pulps having undergone the discoloration-improving treatment, printing papers of the present invention use other raw pulps such as chemical pulps including bleached softwood kraft pulps (NBKPs) or

unbleached kraft pulps (NUKPs), bleached hardwood kraft pulps (LBKPs) or unbleached kraft pulps (LUKPs), etc.; mechanical pulps (MPs) including groundwood pulps (GPs), thermomechanical pulps (TMPs), chemithermomechanical pulps (CTMPs), etc.; and deinked pulps (DIPs) alone or in combination and mixed at any ratio. The pH during the papermaking process may be acidic or neutral or alkaline.

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Printing papers of the present invention can contain

10 fillers. Fillers that can be used include known fillers such as white carbon, talc, kaolin, clay, heavy calcium carbonate, light calcium carbonate, titanium oxide, synthetic resin fillers, etc.

[0037]

If desired, printing papers of the present invention can further contain aluminum sulfate, sizing agents, yield improvers, freeness improvers, colorants, dyes, antifoaming agents or the like.

[0038]

Printing papers of the present invention may or may not be coated with a finishing agent containing no pigment.

Non-coated printing papers of the present invention are desirably coated with a finishing agent based on a water-soluble polymer for the purpose of improving surface

25 strength or sizing performance. Suitable water-soluble polymers include commonly used finishing agents such as starches, oxidized starches, modified starches, carboxymethylcellulose, polyacrylamide, polyvinyl alcohol,

etc. alone or as mixtures thereof. In addition to the water-soluble polymers, the finishing agents can also contain paper strength enhancers designed to improve water resistance or surface strength and external sizing additives designed to provide sizing performance. The finishing agents can be applied with coaters such as two-roll size press coaters, gate roll coaters, blade metering coaters, rod metering coaters, etc., but film transfer coaters such as gate roll coaters are preferable because the finishing agents remain on the paper surface to control the increase in density. The finishing agents are preferably applied in an amount of 0.1 g/m² or more and 3 g/m² or less per side.

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15 Thus, papers with reduced discoloration and reduced brightness change can be obtained by including a pulp having undergone a discoloration improving treatment or performing a discoloration improving treatment on paper after it is prepared. Especially, excellent bulky and 20 light weight papers capable of effectively utilizing forest resources can be obtained because a larger quantity of MP can be included in high brightness papers that were previously difficult to blend with MP due to the problem of discoloration by adding bulkier MP having higher yield than 25 CP after it is selectively subjected to a discoloration improving treatment. Papers of the present invention can be used not only for books, but also for offset printing, relief printing, gravure printing, newspaper printing; and

as electro-photographic papers, or base papers for coated papers, inkjet recording papers, thermosensitive papers, pressure sensitive papers or the like.

EXAMPLES

5 [0040]

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The following examples illustrate the present invention in detail without, however, limiting the scope of the invention thereto. The papers prepared in the examples and comparative examples were tested for their quality under the following items. The evaluation methods are as follows.

-Basis weight: determined according to JIS P 8124:1998.

-Paper thickness and density: determined according to 15 JIS P 8118:1998.

-ISO brightness: determined according to JIS P 8148.

-Fading test: ISO brightness was determined by a xenon lamp weather meter (Suga Test Instruments Co., Ltd.) after irradiating samples with UV light generated from a xenon lamp for 1.0 hour. Test conditions included a black panel temperature of 63°C, humidity of 50%, and an irradiation intensity of 70 W.

[Example 1]

A paper having a basis weight of 46 g/m² was prepared
in an oriented sheet former (Kumagai Riki Kogyo Co.) from a
mixed pulp slurry consisting of 10 parts of BCTMP (1)
(bleached chemithermomechanical pulp available from Millar
Western, ISO brightness 85%, freeness CSF 250 ml) and 90

parts of LBKP (freeness CSF 400 ml) containing 0.9% (based on pulp) of aluminum sulfate, 0.5% of cationized starch as an internal additive (trade name: Cato 3210 available from Nippon Scientific Co., Ltd.), and a blend of light calcium carbonate and talc in a ratio of 6:4 in an amount corresponding to an ash content of 6% in the paper. This sample was immersed in an aqueous 6% sodium borohydride solution (w/v) and then placed on a glass plate. At this point, the impregnant content of sodium borohydride was 18% solids by weight on the basis of the weight of pulp solids. This was irradiated with KrF excimer laser at 40 mJ/cm²/pulse, 5 Hz for 40 minutes. After completion of laser irradiation, the sample was washed with water and dried in a cylinder dryer, and then evaluated by the fading test.

[Example 2]

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A similar procedure as in Example 1 was carried out, except that a mixture of 30 parts of BCTMP (1) and 70 parts of LBKP was used.

20 [Example 3]

A similar procedure as in Example 2 was carried out except that BCTMP (1) was replaced by BCTMP (2) (Millar Western, ISO brightness 70%, softwood 25%, aspen 75%).

[Example 4]

A pulp dispersion of BCTMP (1) (Millar Western, ISO brightness 85%, freeness 250 ml C.S.F.) broken at a concentration of 2% was combined with 20% of sodium borohydride (w/v) based on pulp and then introduced into a

transparent plastic bag and irradiated with UV light from a low-pressure mercury lamp (Photoscience Japan Corporation, AY-1, 244 w/m²) for 12 hours and washed with water. Then, a paper having a basis weight of 46 g/m² was prepared in an oriented sheet former (Kumagai Riki Kogyo Co.) from a mixed pulp slurry consisting of 10 parts of this BCTMP (1) irradiated with UV light and 90 parts of LBKP (freeness 4000 ml C.S.F.) containing 0.9% (based on pulp) of aluminum sulfate, 0.5% of cationized starch as an internal additive and light calcium carbonate in an amount corresponding to an ash content of 6% in the paper. This sample was evaluated by the fading test.

[Example 5]

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A similar procedure as in Example 4 was carried out

15 except that a mixture of 30 parts of BCTMP (1) irradiated
with UV light and 70 parts of LBKP was used.

[Example 6]

The same procedure as in Example 4 was performed except that BCTMP (2) (Millar Western, ISO brightness 70%, softwood 25%, aspen 75%) irradiated with UV light was used in the pulp mixture.

[Comparative example 1]

A paper was prepared in an oriented sheet former in the same manner as in Example 1 from 100 parts of LBKP with the internal chemicals and fillers without using BCTMP.

This sample was evaluated by the fading test.

[Comparative example 2]

The sample prepared in the oriented sheet former in

Example 1 was directly evaluated by the fading test.

[Comparative example 3]

The sample prepared in the oriented sheet former in Example 2 was directly evaluated by the fading test.

5 [Comparative example 4]

The sample prepared in the oriented sheet former in Example 3 was directly evaluated by the fading test.

10 Table 1

[0041]

| | Brightness in fading test | | Basis weight | Thickness | Density |
|-----------------------|---------------------------|--------|-----------------|-----------|---------|
| | 0 hour | 1 hour | (g/m²) | (µm) | (g/cm³) |
| Comparative example 1 | 79.7 | 78.5 | 45.8 | 69 | 0.66 |
| Comparative example 2 | 78.3 | 72.1 | 45.8 | 75 | 0.61 |
| Comparative example 3 | 77.4 | 64.7 | 46.2 | 79 | 0.58 |
| Comparative example 4 | 70.3 | 59.6 | 45.7 | 80 | 0.57 |
| Example 1 | 78.3 | 76.3 | 45.8 | 75 | 0.61 |
| Example 2 | 78 | 75.6 | 46.2 | 79 | 0.58 |
| Example 3 | 76.4 | 73.7 | 45.7 | 80 | 0.57 |
| Example 4 | 78.3 | 76.3 | 45.8 | 75 | 0.61 |
| Example 5 | 78.0 | 75.6 | 46.2 | 79 | 0.58 |
| Example 6 | 76.8 | 73.7 | 45.7 | 80 | 0.57 |

It is apparent from Table 1 that the papers having undergone an anti-fading treatment show a great reduction in brightness loss, i.e., they become more resistant to

Examples 3 and 6 using MP having lower discoloration. brightness improved in brightness by the anti-fading treatment to a level higher than the brightness of Comparative example 4 using the same MP without anti-fading 5 treatment and comparable to the brightness of Comparative example 1 using LBKP alone without MP. The presence of bulky MP helps add bulk and greatly reduces the density as compared with Comparative example 1, leading to weight reduction because the same paper thickness is achieved at a lower basis weight. Comparative examples 2-4 also show 10 weight reduction as compared with Comparative example 1 using no MP, but suffer from severe discoloration. contrast, the examples of the present invention are wellsuited for making book papers because of greatly reduced discoloration and a combination of weight reduction by 15 including MP and the anti-fading treatment, and they also contribute to resource saving because they can contain MP having high yield.